

Soils for salad and sweet box

Cultivating success from the ground up



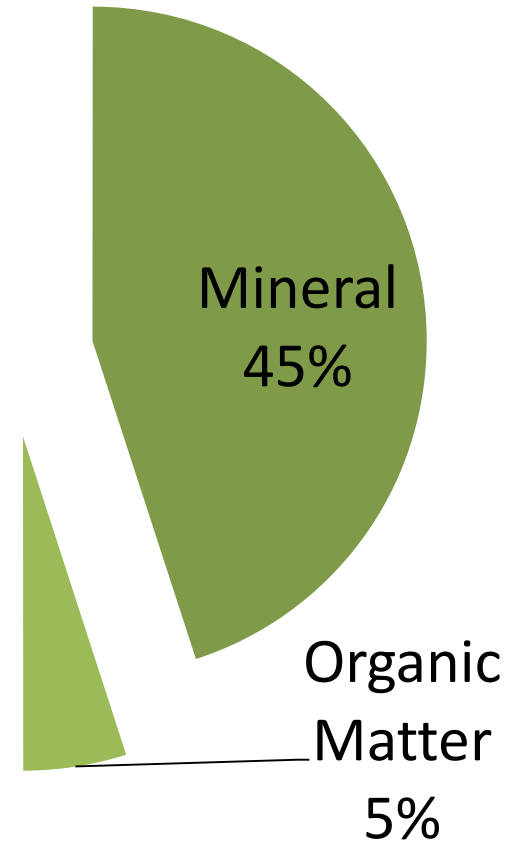
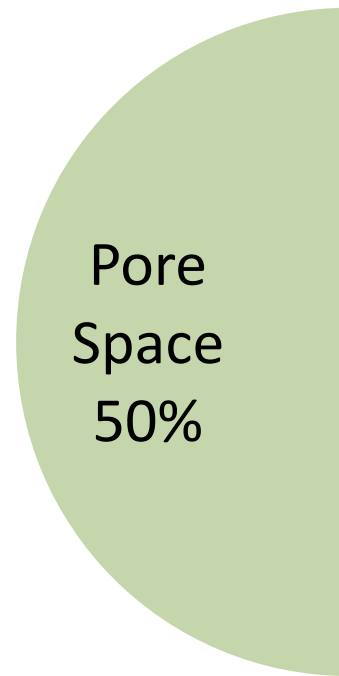
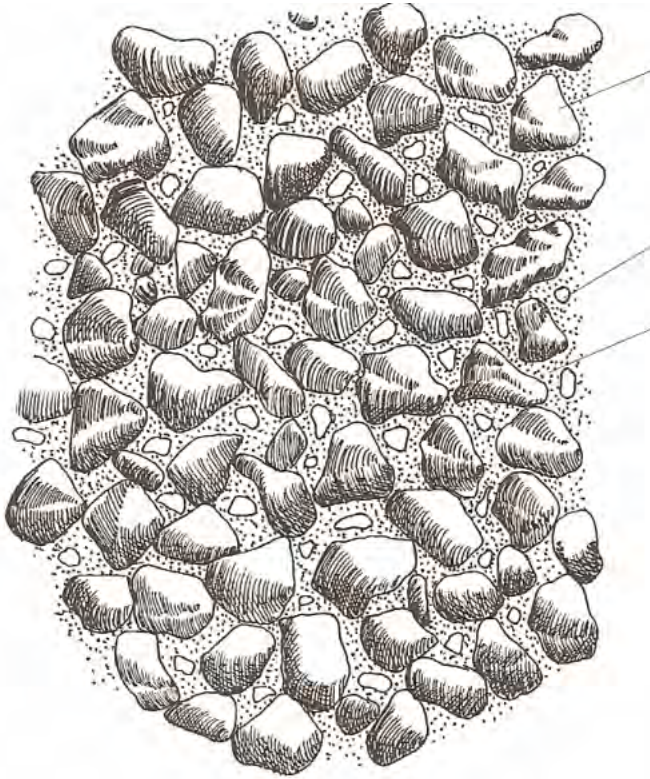
Kate Kurtz

Soil scientist, biosolids project manager, King County

- What is soil?
- How is it made?
- Texture
- Structure
- Biology
- Fertility/nutrients



What is soil?



How is soil made?

Parent Material: rocks or alluvium

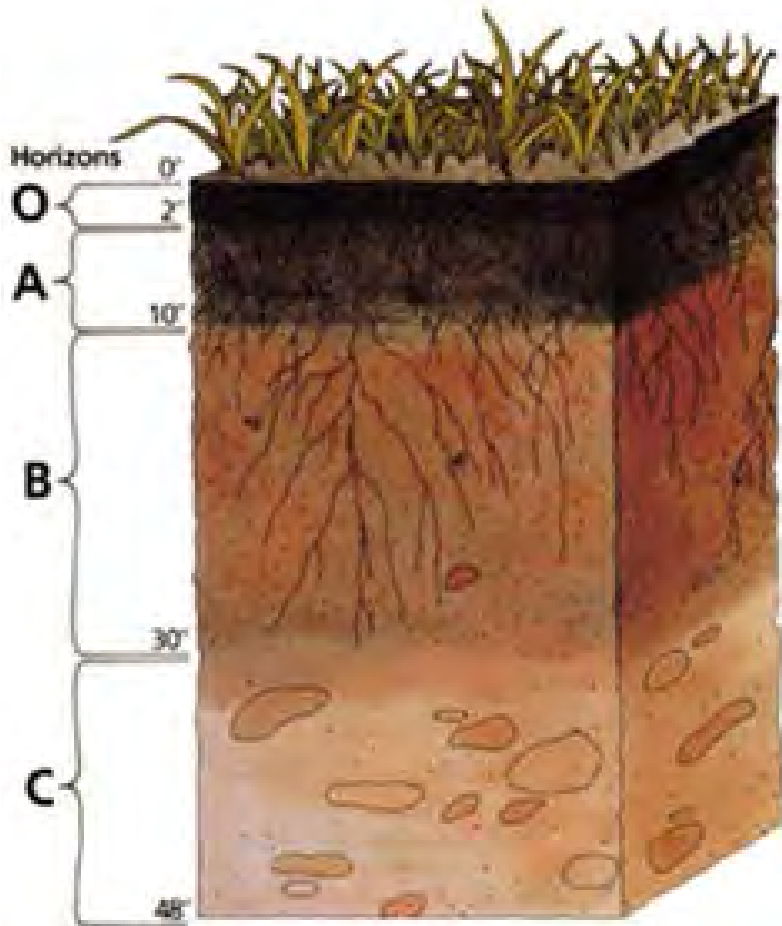
Climate: Temperature, moisture, weather, and seasonal distribution.

Topography: Slope, aspect, shape.

Living organisms: animals and plants are often source of organic matter and many nutrients.

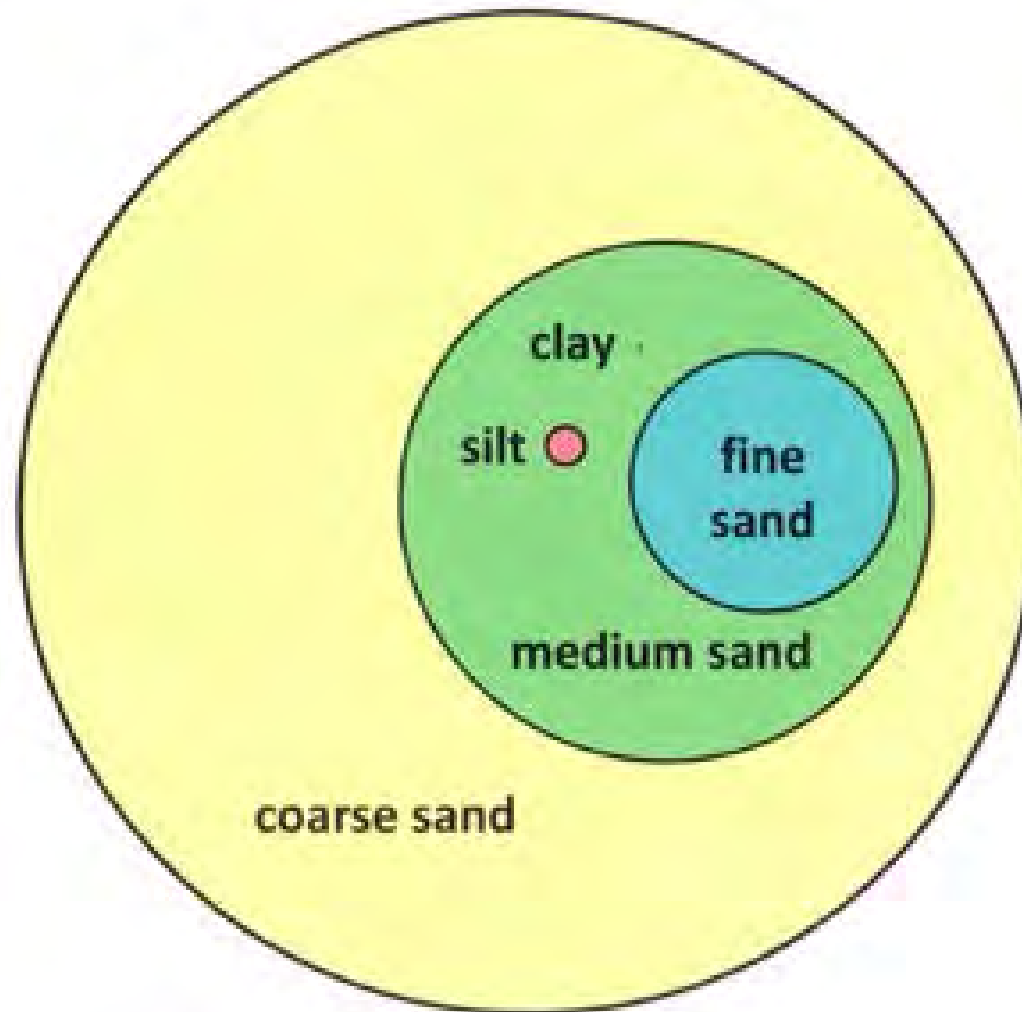
Time: allows all the preceding factors to work. Over eons rocks breakdown into clay particles and leach. Young soils tend to have less clay and are more fertile.

What does soil look like?



Soil Texture

Relative amounts of sand, silt, and clay



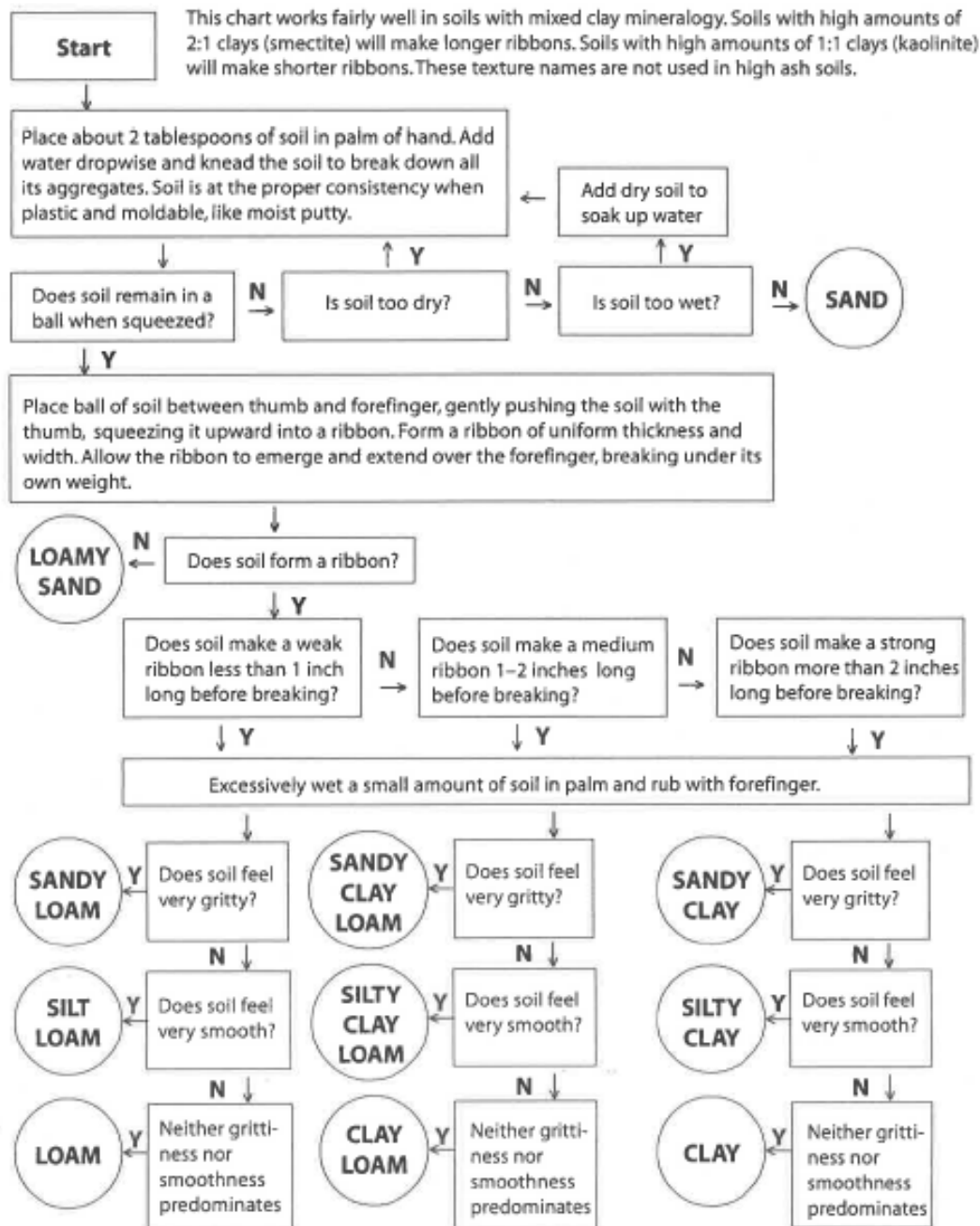
Soil texture

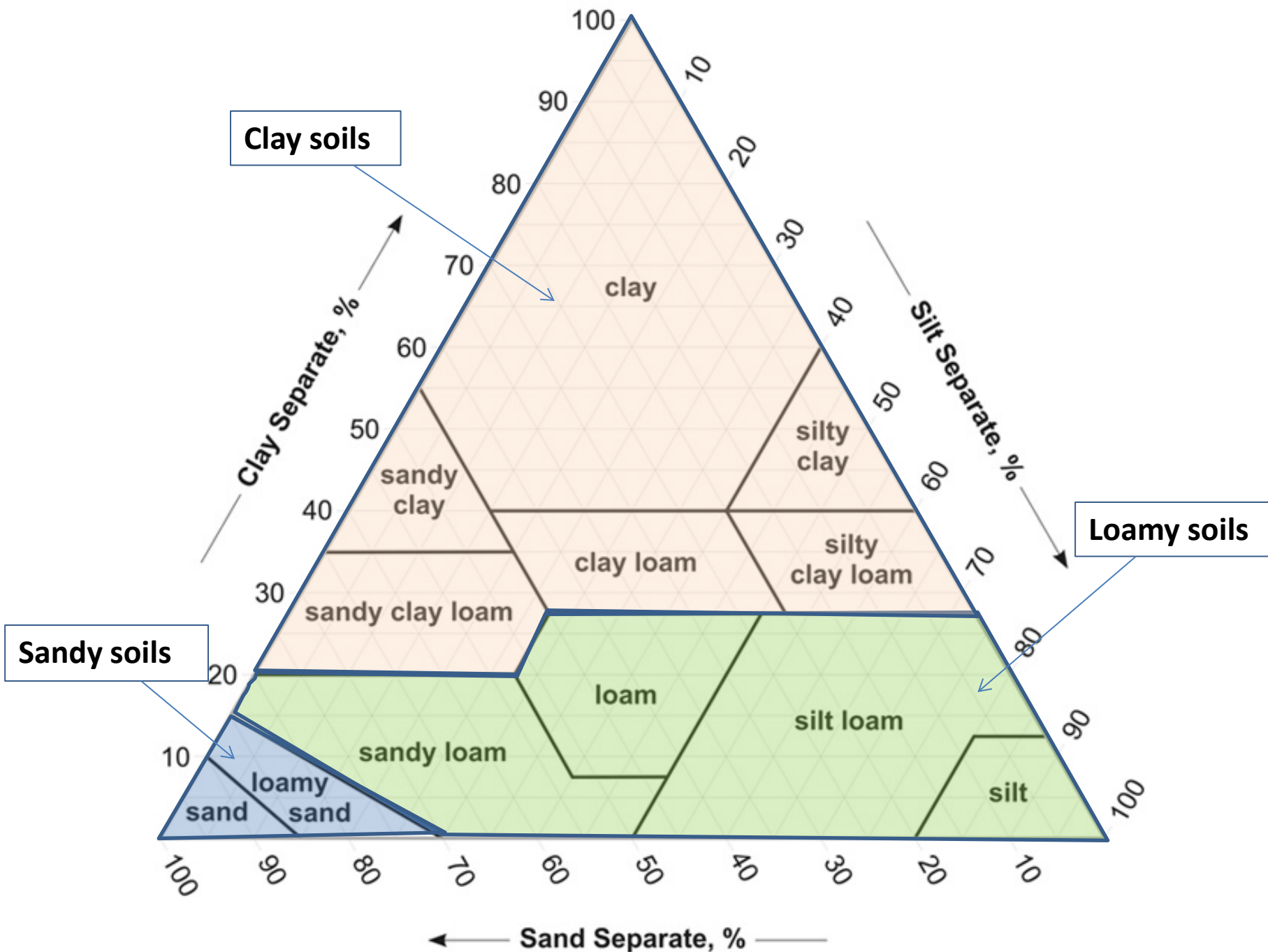
- Sand feels gritty
- Silt feels floury when dry but greasy when wet
- Clay feels sticky and is easily molded into shapes

HOW TEXTURE AFFECTS SOIL PROPERTIES

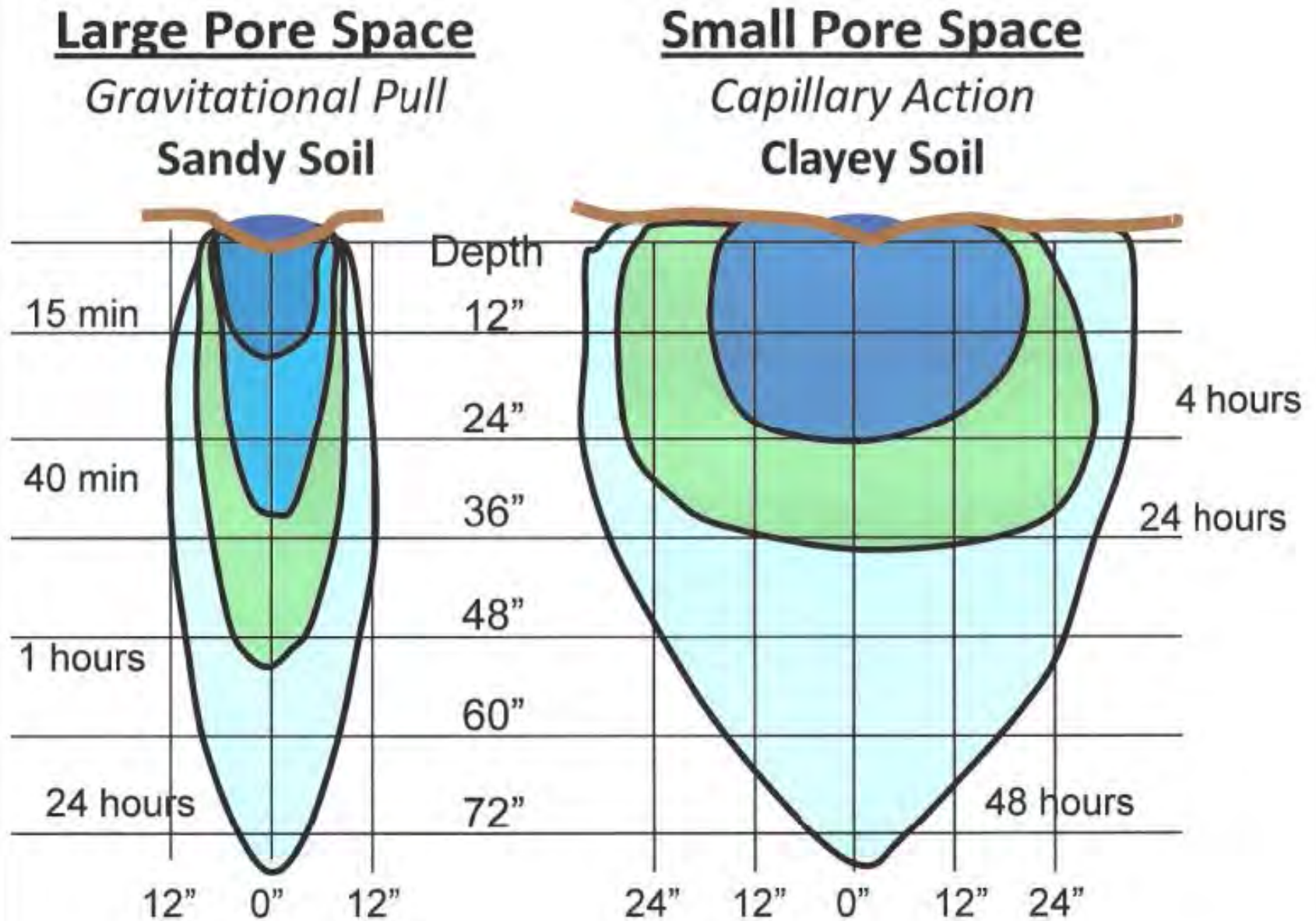
Texture	Aeration/ Porosity	Ease of Water Infiltration	Ability to Hold Nutrients	Water-Holding Capacity	Ease of Working
Loam	medium	medium	medium	medium	medium
Clay	poor	poor	excellent	good	poor
Silt	medium	medium	medium	medium	medium
Sand	excellent	good	very poor	very poor	good

Soil Texture Decision Chart





How texture affects drainage



Drainage and water holding capacity



- Macropores control infiltration and drainage
- Small pores control water holding capacity
- Tiniest of pores hold water unavailable to plants

Soil structure: crumbs, clods, or mud pies? (under cultivation)

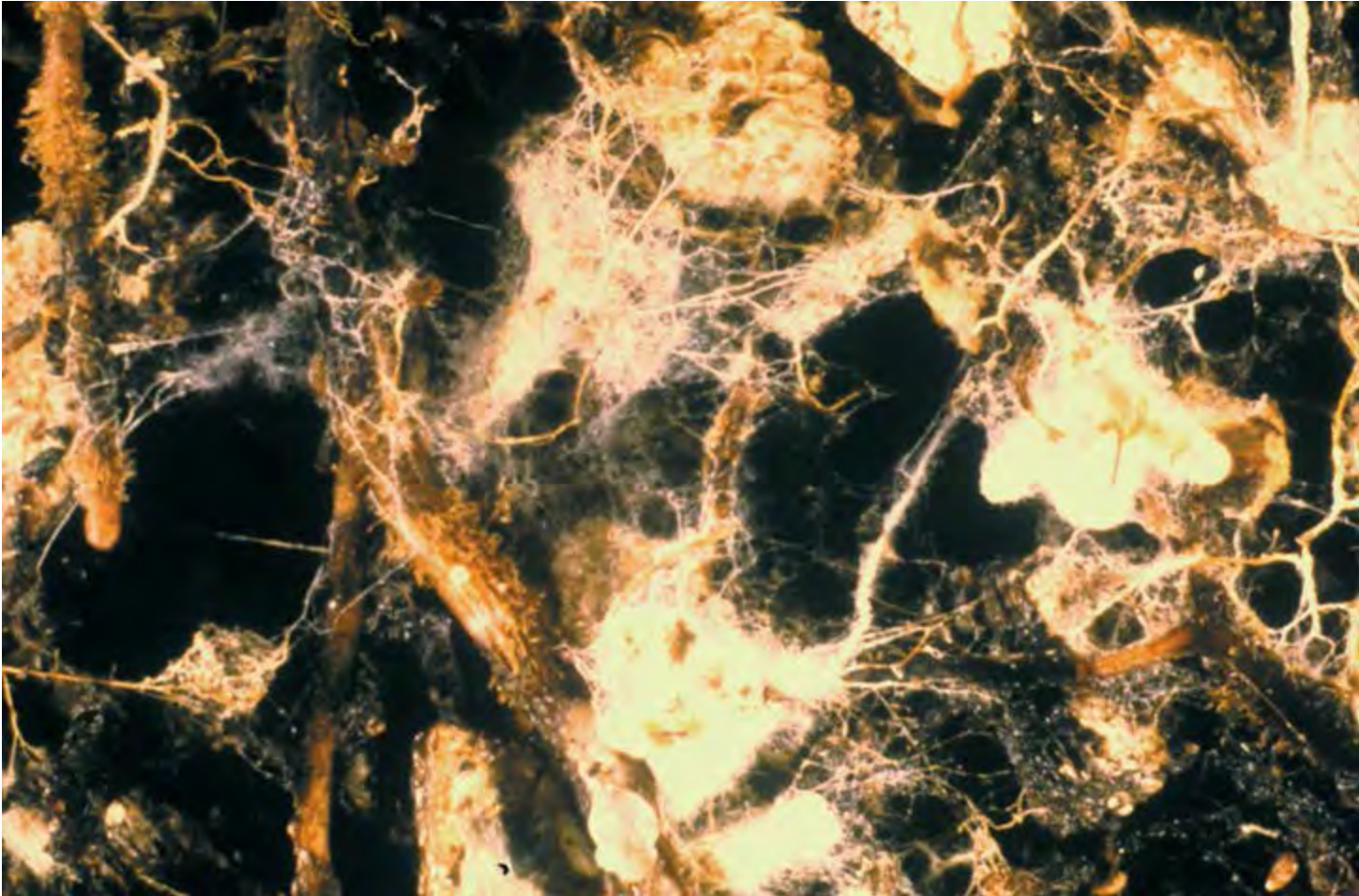


Compaction and how to ruin your soil's structure



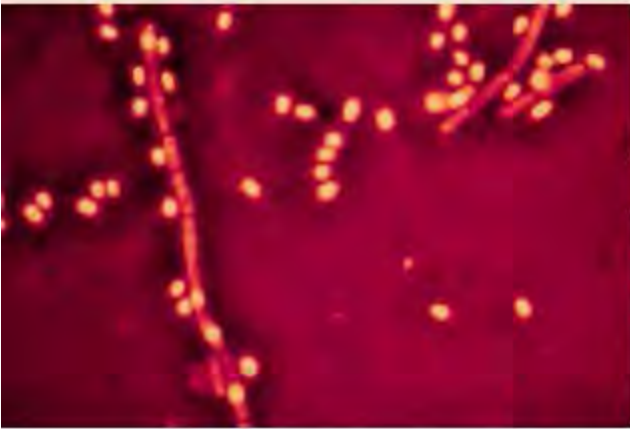
Improve soil structure

Organic Matter and Soil Microbes



Helps soil structure by tangling and gluing soil particles together

Soil Organisms



Bacteria, fungi, actinomycetes, protozoa,
nematodes, arthropods, earthworms

Roles of Soil Organisms

- Residue decomposition
- Nutrient cycling
- Aggregation and porosity
- Contaminant breakdown
- Nitrogen fixation
- Enhance root function
- Pathogens
- Predators



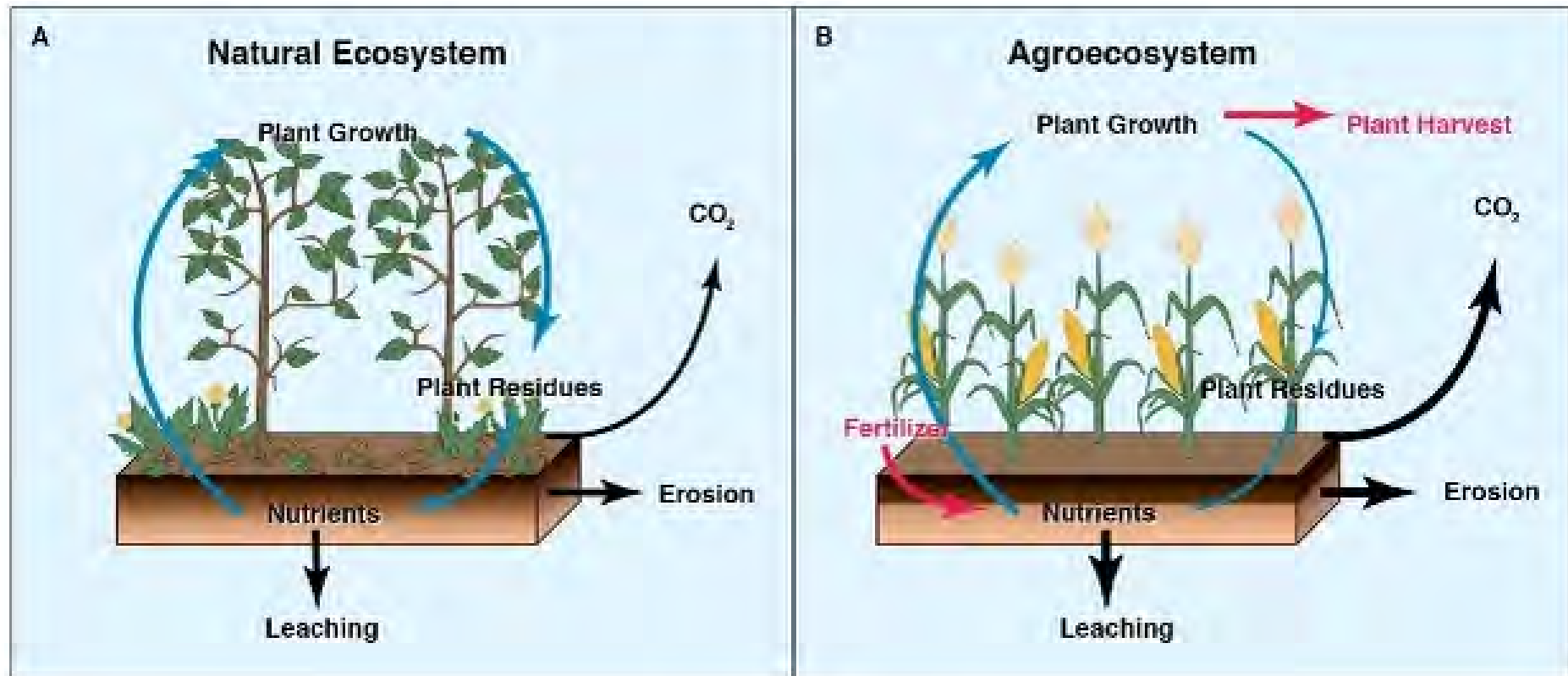
Improving drainage



Improving drainage



Fertility/nutrients



Plant Nutrients

Macronutrients

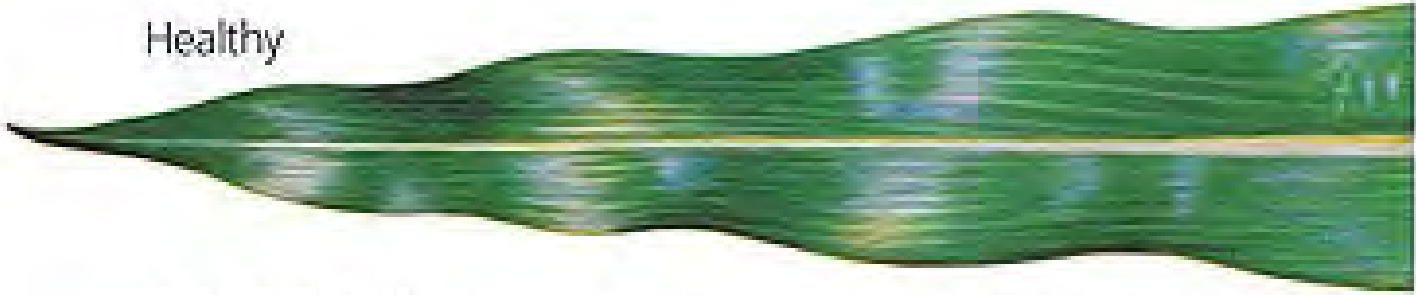
- Nitrogen
- Phosphorus
- Potassium
- Calcium
- Magnesium
- Sulfur

Micronutrients

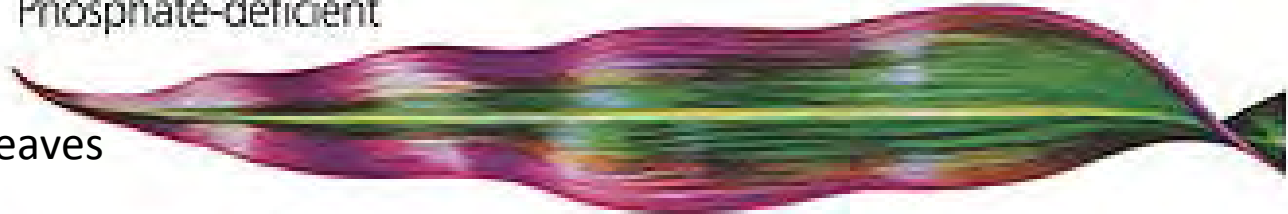
- Boron
- Iron
- Manganese
- Zinc
- Copper
- Chloride
- Molybdenum

Nutrient deficiency symptoms

Healthy



Phosphate-deficient



P-deficient:
usually see purple leaves

Potassium-deficient



K-deficient:
usually see scorched,
curled leaves

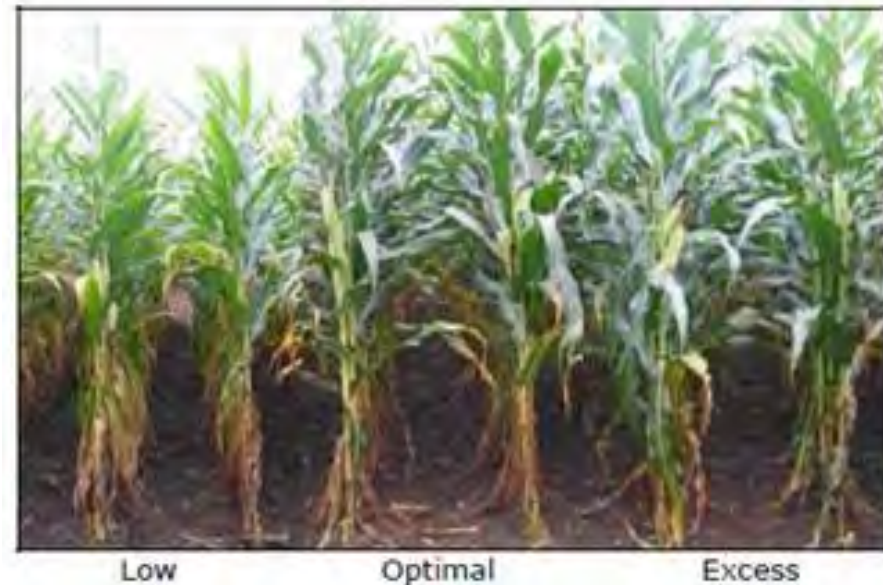
Nitrogen-deficient



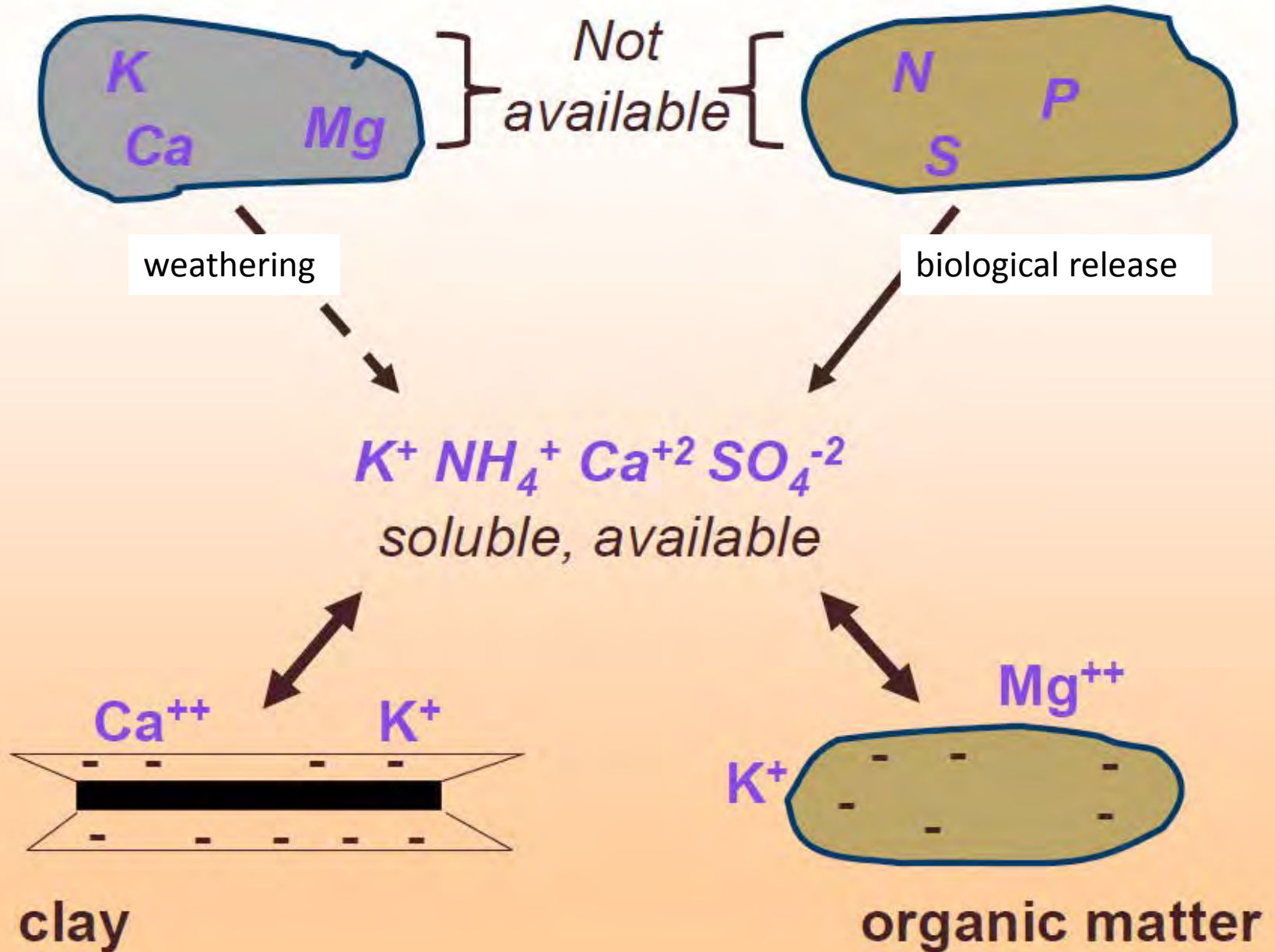
N-deficient:
usually see pale,
chlorotic leaves

Excess nutrients

- Nitrogen:
 - Groundwater contamination
 - Plant health, fruit yield and quality
 - All leaves no fruit
- Phosphorus:
 - Surface water contamination
- Boron
 - Toxicity
- Too much of everything
 - Reduced yield and vigor
 - Problems taking up water



How do nutrients become available?



Comparing Fertilizer Sources

Organic

- Little or no processing
- Low nutrient analysis
- Usually slow release
- Often unknown concentrations
- Usually a source of organic matter



Synthetic

- Industrial process with potentially major environmental impacts (huge carbon emissions, water contamination)
- Usually fast release
- Known concentrations
- No organic matter



Comparing fertilizer Sources

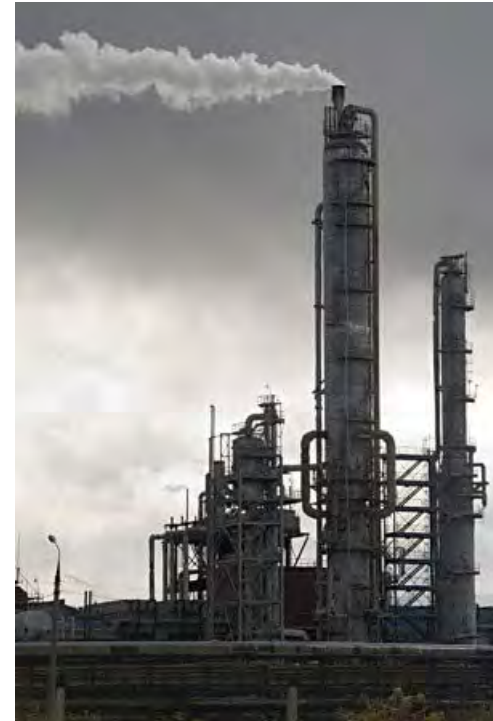
Organic

- Manure
- Biosolids
- Bone meal
- Wood ash
- Blood meal
- Seed meal



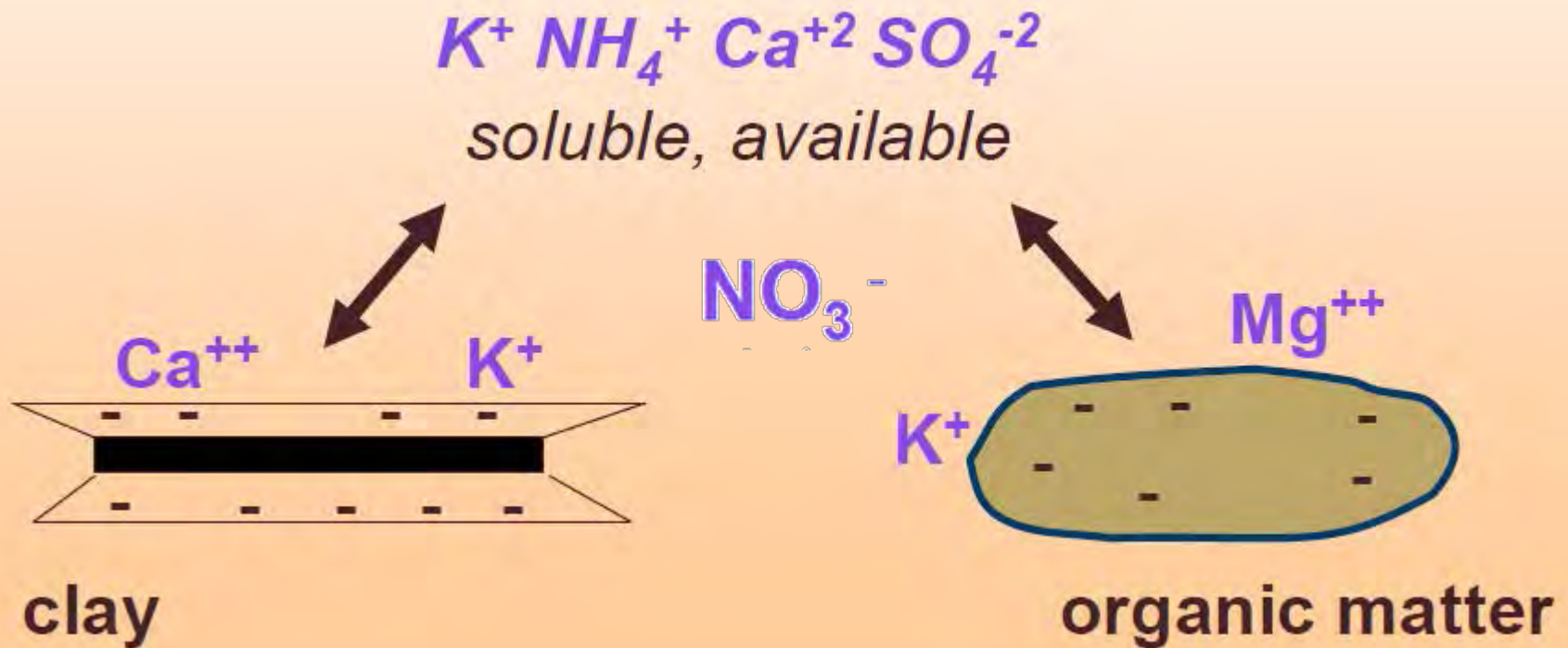
Synthetic

- N – air
- P – mined
- K - mined



Nutrient Uptake

The forms of nutrients taken up by plants are the same regardless of the nutrient source.



Fertilizer Labels



5 — 10 — 10
% Nitrogen — % Phosphate — % Potash

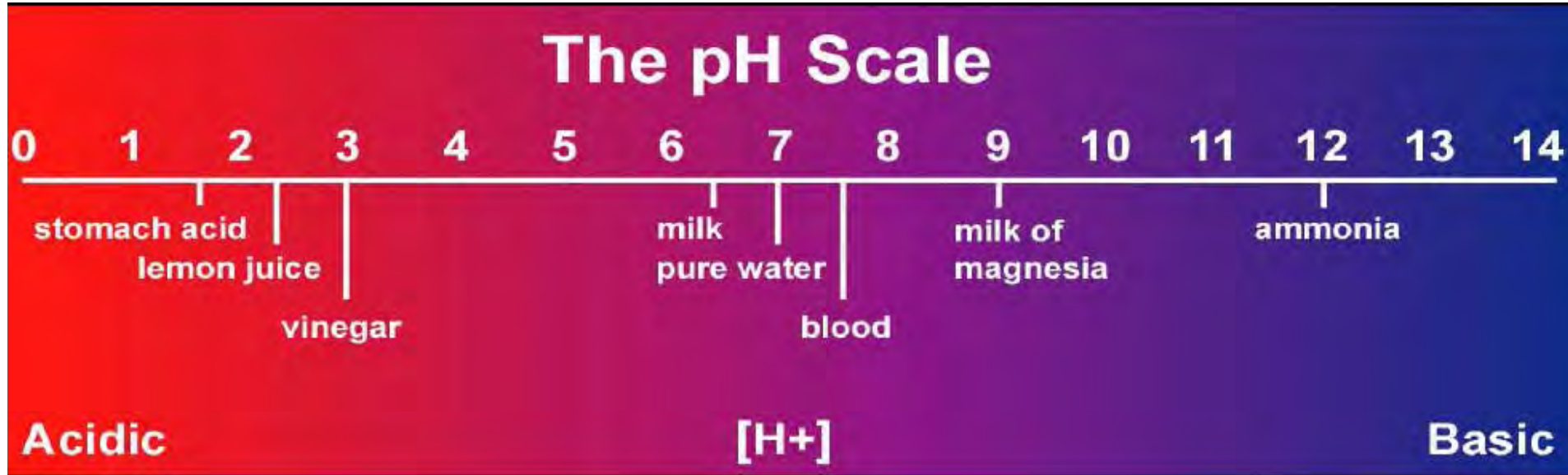
Phosphate = units of P

1 lb P = 2.3 lb phosphate (P_2O_5)

Potash = units of K

1 lb K = 1.2 lb potash (K_2O)

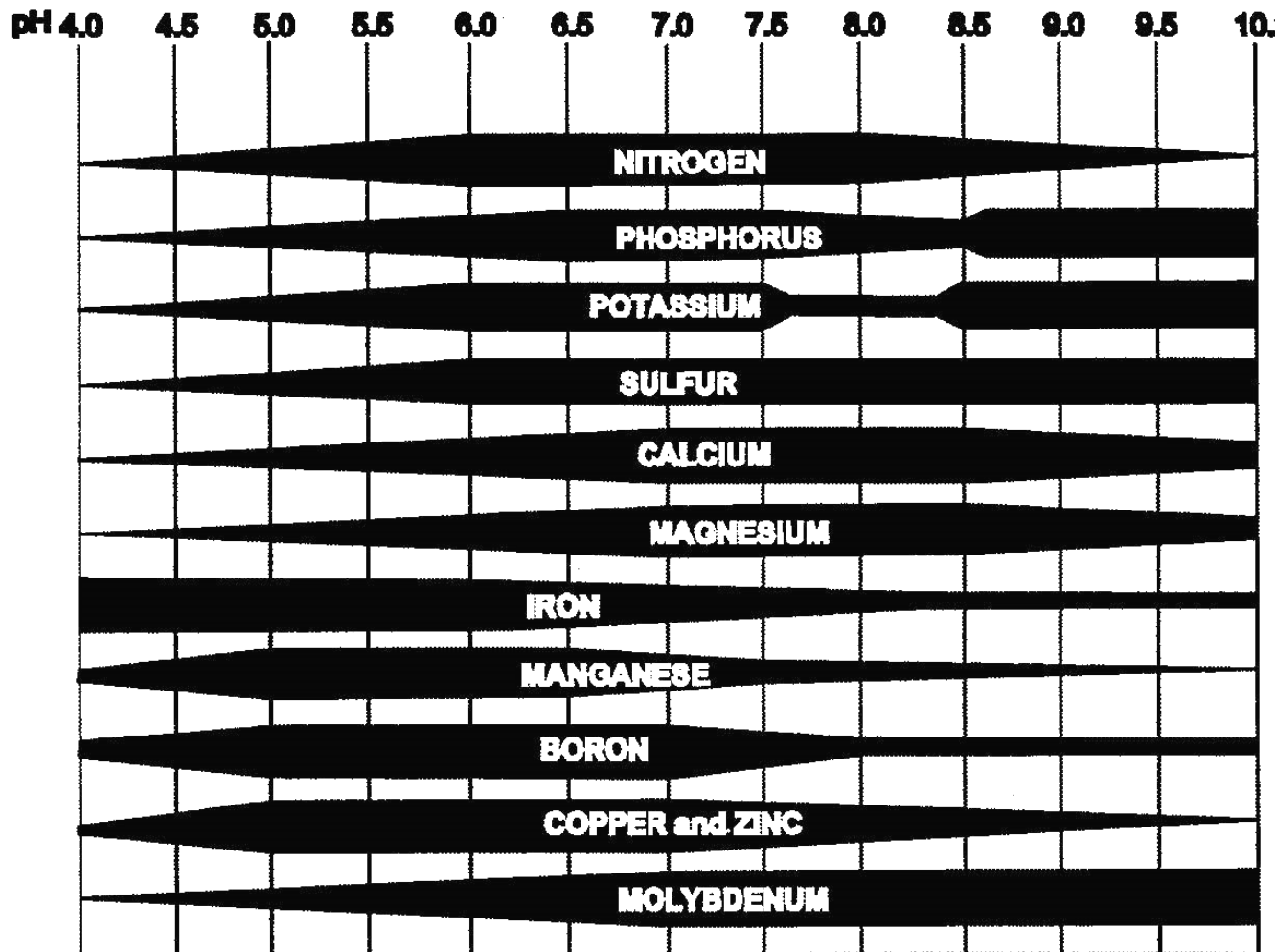
Soil pH



- Indicated relative acidity or alkalinity
- pH 7 = neutral
- pH less than 7 = acid
- pH more than 7 = alkaline (basic)
- Logarithmic scale

Soil pH

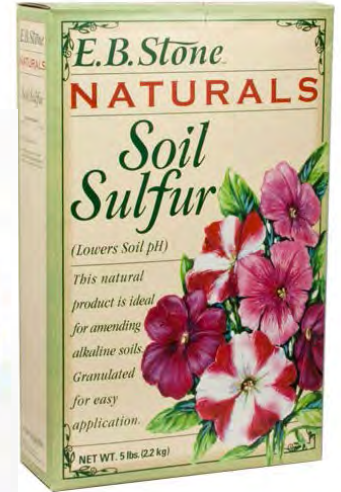
NUTRIENT AVAILABILITY AT DIFFERENT pH VALUES. MAXIMUM AVAILABILITY IS INDICATED BY WIDEST PART OF BAR



6 - 7.5 is a sweet spot for most plants

Adjusting pH

- Organic matter tends to buffer pH around neutral
- To lower pH add sulfur
- To raise pH add lime



Agricultural Lime



Dolomite Lime



Oyster Shell Flour



Gypsum

Testing soil



- Sample defined area
- Take 10 or more subsamples (0-8" deep)



- Dry and mix well








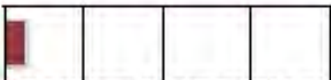
- Subsample mixture and fill sample bag





Name: KRISTIN COVEY
Date Received: 1/9/2017

Sample ID: GARDEN
Kit Number:

Test Results	Interpretation	Fertility Recommendations
● pH 6.5	 <p>Acidic Neutral Basic</p>	
● Salts 0.44	 <p>Low Medium High</p>	
● Organic Matter (%) 5.1	 <p>3 6 9</p>	
● Ammonium-N (ppm) 4.2	 <p>Deficient Sufficient High</p>	● 0.0 lbs of N per 1,000 ft ²
● Nitrate-N (ppm) 12.2	<p>Ammonium + Nitrate + OM</p>  <p>Deficient Sufficient High</p>	● 0.0 lbs of P ₂ O ₅ per 1,000 ft ²
● Phosphorus 271	 <p>Deficient Sufficient High</p>	● 2.5 lbs of K ₂ O per 1,000 ft ²
● Potassium 52		

What does this test mean

Need more assistance?

Visit www.soiltestlab.com

pH: The pH scale is from 0-14 with 7 being neutral. If your pH is less than 7 the soil is acidic, if greater than 7 it is basic. To lower pH elemental sulfur can be added, to raise pH lime can be added. Most plants can grow in a wide range of soil pH, however, optimal pH is plant dependent.

**SOILTEST FARM CONSULTANTS - 11**

2925 DRIGGS DR

Moses Lake, WA 98837

Laboratory #: S17-00161

Date Received: 1/9/2017

Grower: KRISTIN COVEY

Sampled By:

Field: GARDEN

Customer Account #:

Customer Sample ID:

Soil Test Results

Phosphorus	Bray	mg/kg	271
Potassium	NH4OAc	mg/kg	44
Boron	DTPA	mg/kg	0.17
Zinc	DTPA	mg/kg	11.5
Calcium	NH4OAc	meq/100g	6.1
Magnesium	NH4OAc	meq/100g	2.2
Sodium	NH4OAc	meq/100g	0.33
Lime Req		Tons/Acre	0.0
Buffer pH	SMP		6.7
Total Bases	NH4OAc	meq/100g	8.7

pH 1:1 6.5 CaCl2 pH 5.8

E.C. 1:1 m.mhos/cm 0.17

Est Sat Paste E.C. m.mhos/cm 0.44

EffervescenceLbs/Acre

Ammonium - N mg/kg 4.2 13

Organic Matter W.B. % 5.1 ENR: 102

Depth	Nitrate-N	Sulfate-S	Moisture
inches	mg/kg	lbs/acre	mg/kg
			inches
0 - 12	12.2	39	13
Totals	12.2	39	13

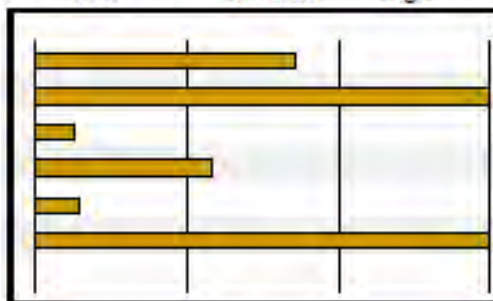
Sum of Tested N: 154 lbs/acre N

Interpretation Guide**Fertilizer recommendations for**

of GARDEN after

Low Medium High

Nitrogen 154 lbs/acre
 Phosphorus 271 mg/kg
 Potassium 44 mg/kg
 Sulfur 13 mg/kg
 Boron 0.17 mg/kg
 Zinc 11.5 mg/kg
 Manganese mg/kg



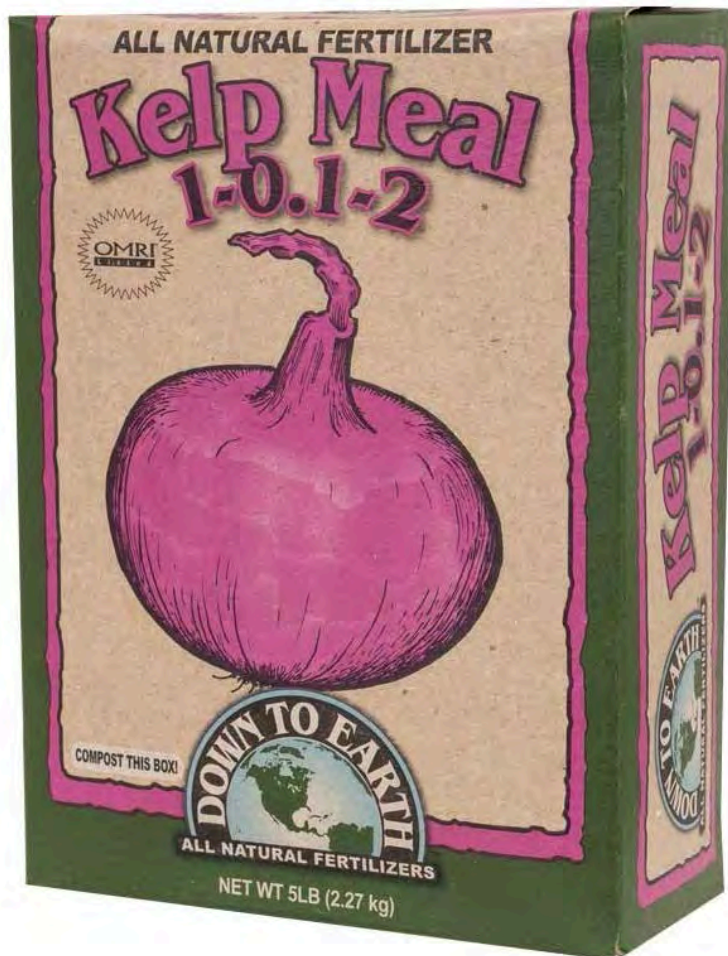
0 lbs/acre of Nitrogen
 0 lbs/acre of P2O5
 115 lbs/acre of K2O
 0 lbs/acre of Sulfur
 1 lbs/acre of Boron
 0 lbs/acre of Zinc
 0 lbs/acre of Mn

We make every effort to provide an accurate analysis of your sample. For reasonable cause we will repeat tests, but because of factors beyond our control in sampling procedures and the inherent variability of soil, our liability is limited to the price of the test. Recommendations are to be used as general guides and should be modified for specific field conditions and situations. Note: "u" indicates that the element was analyzed for but not detected.

This is your Invoice #: S17-00161 Account # 101100 Reviewed by: KEB

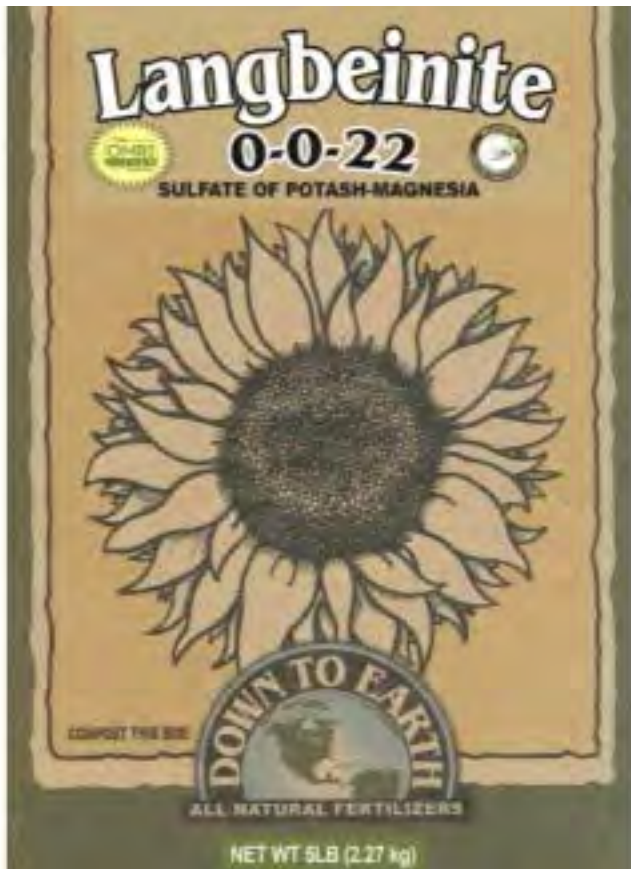
List Cost: \$30.00

Recommendation:
2.5 lbs K_2O / 1000 ft²



- This fertilizer is 2% K_2O
- $0.02 * X = 2.5\text{lbs}$
- $X = 125\text{lbs}$ of fertilizer / 1000 ft²
 - (that would be 25 5lb boxes)
- She has approximately 200 ft² of garden space (1/5 of 1000)
- So she needs 25lbs of this fertilizer for her garden. That's the same as 5, 5lb boxes of this fertilizer spread evenly across her garden area.

Recommendation:
 $2.5 \text{ lbs } K_2O / 1000 \text{ ft}^2$



- This fertilizer is 22% K_2O
- $0.22 * X = 2.5 \text{ lbs}$
- $X = 11.4 \text{ lbs of fertilizer} / 1000 \text{ ft}^2$
 - (that would be a little more than 2 boxes)
- She has approximately 200 ft^2 of garden space ($1/5$ of 1000)
- So she needs 2.3lbs of this fertilizer for her garden. That's the same as about half of 1 5lb box of this fertilizer spread evenly across her garden area.

Carbon:Nitrogen Ratio

- Ranges from $<5:1$ to $>500:1$ in organic materials
- Low C:N supplies N to plants
- High C:N ties up N by biological immobilization

Types of Organic Amendments

Hot stuff – C:N < 10:1

Cool stuff – C:N 15:1 to 25:1

Woody stuff – C:N > 30:1

Hot Stuff

- Poultry manure
- Seed meals
- Fish and feather meals
- Fresh grass clippings
- Fresh rabbit manure
- Dewatered or dried biosolids



Hot Stuff C:N < 10:1

- Rapid N availability
- Use as a fertilizer
- Over application leads to excess nutrient levels in soil -- potentially harming crop and water quality.

Cool stuff

- Compost (yard debris, most manures, biosolids)
- Fresh materials:
 - Yard debris
 - Cover crop residues
 - Dairy manure solids
 - Coffee grounds



Cool stuff, C:N 15:1 to 25:1

- Slow N availability
- Can add large amounts without risk of over-fertilization
- Use as a soil amendment
- N immobilization (tie-up) likely with fresh materials in first few weeks after application
- Compost organic matter lasts a long time in soil

Woody stuff

- Straw
- Sawdust
- Paper waste
- Horse manure rich in bedding



Woody stuff, C:N > 30:1

- N immobilization
- Need to add N along with organic amendment
- Use as mulch or bulking agent for compost

Organic Application Guidelines

(middle-of-the road estimates)

- Poultry manure: 3-5 gal/100 sq. ft.
- Rabbit manure: 20-40 gal/100 sq. ft.
- Horse manure, dairy solids: 1 inch
- Compost: 1 inch per year in annual bed
- Compost: Up to 30 to 50% by volume when making raised beds.
- Seed and animal by-product meals: Use like inorganic fertilizers

Green Manure or Cover Crops

- Grow your own OM
- Legumes will provide N
- Cover crop during the winter
- Early planting is best
- Soil incorporation takes work



Common Winter Cover Crops



Cereal Rye



Annual Ryegrass



Hairy Vetch



Crimson Clover

Compost

- Home or municipal
- Good source of OM
- Nutrient value depends on compost type



The soil is the great connector of our lives,
the source and the destination of all
- Wendell Berry

Kate Kurtz

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